Sl.No:M21196 Course Code: BP104T

VINAYAKA MISSION'S RESEARCH FOUNDATION (Deemed to be University), SALEM

B.PHARM. DEGREE EXAMINATION – October 2021 First Semester

PHARMACEUTICAL INORGANIC CHEMISTRY - II

Time: Three hours

Maximum: 75 marks

I. Write essays on any **TWO** questions:

 $(2 \times 10 = 20)$

- 1. Define radioactivity. How to measure the radioactivity and explain the storage condition and precaution to be followed by handling radioactive substances.
- 2. Classify antacid and write note on acid neutralizing capacity of aluminium hydroxide gel. Give the preparation and uses of any one antacid.
- 3. Give the principle and reaction involved in the limit test for arsenic with neat diagram of apparatus used for it.

II. Write short answers on any **SEVEN** questions:

 $(7 \times 5 = 35)$

- 4. Write the preparation assay and uses of calcium gluconate.
- 5. What are antidotes? Give the preparation properties assay and uses of sodium thiosulphate.
- 6. Define Pharmacopeia. Short note on IP.
- 7. Describe briefly about physiological acid base balance.
- 8. Write about preparation, properties, assay and uses of ferrous sulphate.
- 9. Explain the methods for adjusting isotonicity.
- 10. Write about preparation, properties, assay and uses of H_2O_2 .
- 11. Write a note on oral antiseptics and astringent with example.
- 12. Write a note on emetics. Write preparation, assay and uses of copper sulphate.

III. Write short notes on:

 $(10 \times 2 = 20)$

- 13. Define expectorants and given an example.
- 14. Define Monograph.
- 15. Write a note on alum.
- 16. Composition of ORS.
- 17. Chlorinated lime.
- 18. Osmotic laxative.
- 19. Role of fluoride in dental cavies.
- 20. Define cathertics and give an example.
- 21. Define intercellular compartment.
- 22. Define denttrifrices.